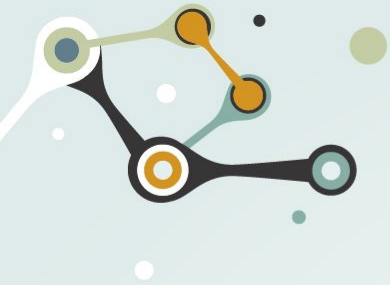




# Self-Made Device to Measure Optical Density of Solutions

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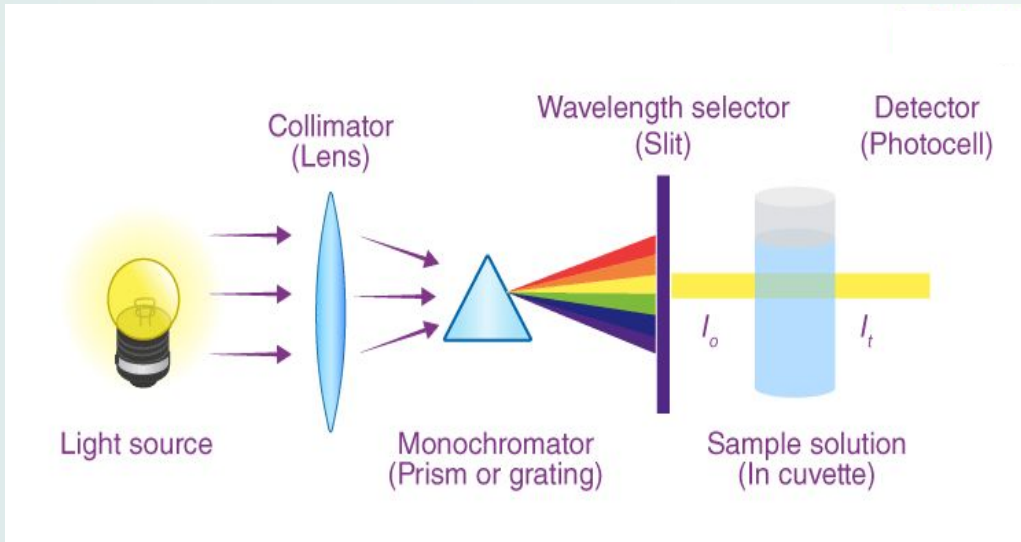
# Rationale of My Design

- I had a lot of fun when I play and compete in all kinds of Lego EV3 Robotics Tournaments and gradually arouse my interest in **Electric Engineering**.
- **Spectrometer with LabQuest** was a tool I used frequently in the past, then I was attempting to create a simple device that may achieve the same goal.
- The process helped me learned a lot in both **hard/software** and gain knowledge about the **optical knowledge** laid behind data.



# The Concept of Optical Density

**Optical density (OD):** is a measure of how much light is absorbed or scattered by a solution.



## Objective of My design:

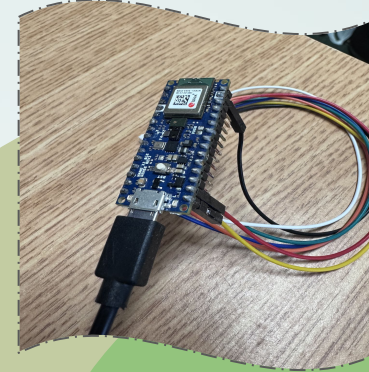
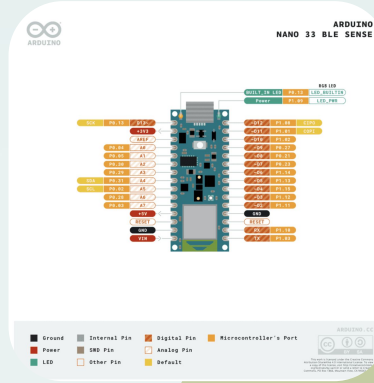
- Use of simple, accessible materials
- Mimic the function of Spectrometer
- Easy to build and conduct experiment
- Explain and learn from the experimental testing result
- What may be improve or fixed?

# Material List

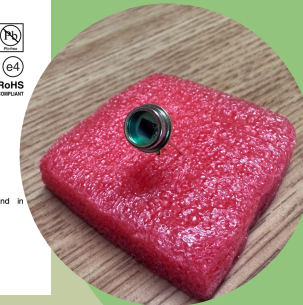
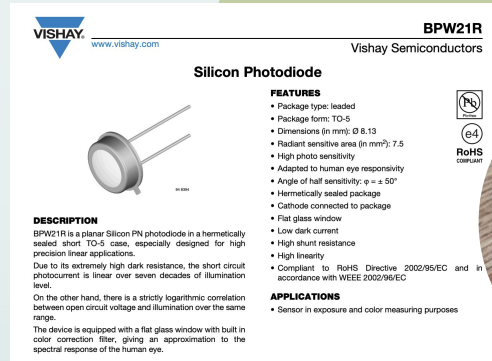
## 1. Customized Acrylic Block



## 2. Microprocessor- Arduino Nano 33 BLE



## 3. Photodiode- BPW21R (Max565nm / 100°)



# Experimental Setup

## Light Sources: 10mm LEDs of different colors

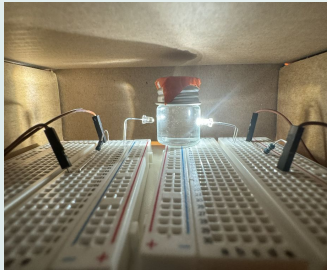
Red (620–630 nm)

Orange (600–605 nm)

Yellow (588–595 nm)

Green (520–525 nm)

Blue (460–465 nm)

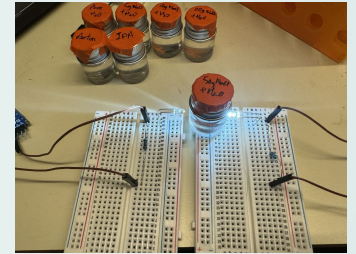


01

02

## Solutions Tested:

1. Pure Water (Baseline)
2. 15% Saline Solution (NaCl)
3. Ethanol
4. Acetone



03

**Sample Holder:** Customized Acrylic Block

**Microcontroller:** Arduino Nano 33 BLE

**Photodiode:** BPW21R (565 nm)

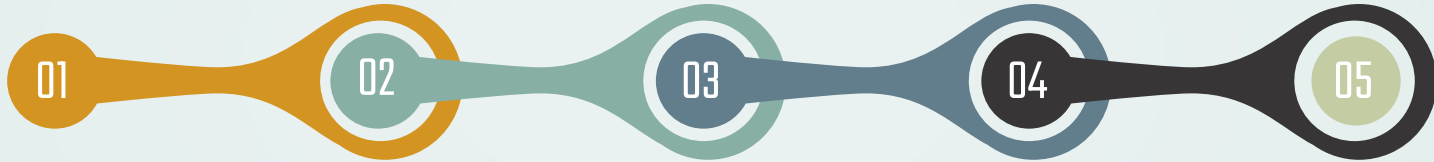
# Procedure of Making it

## Assemble it up

Make sure the circuits and signal pinouts are correctly connected. Check for signal fluctuation

## Sample Testing

Switch the Pure Water with other test solutions.



## Preparation

Obtain all the dupont wires and mini-board, and solution container (cuvette would dworks)

## Calibration Step

Utilize Pure Water to create a baseline value for later data comparison.

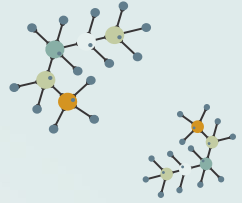
## Data Analysis

Compare the data and analyze the potential reasoning behind the patterns shown



Different substances will exhibit varying absorbance and scattering properties.

# *Hypothesis*



Pure liquids (water, ethanol, acetone) affect light transmission primarily through absorption.



Shorter wavelengths (blue, green) experience more scattering, while longer wavelengths (yellow, orange, red) are less affected.

## **Control Variables:**

1. The matt surface on acrylic block due to CNC process
2. Minor noise and signal disturbance on data
3. Potential light sca

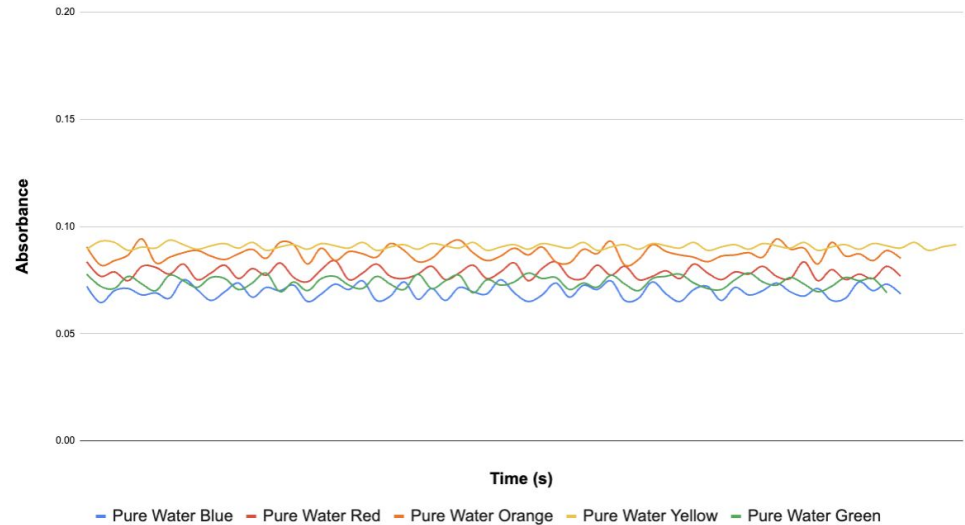


# Data Analysis (Pure Water/Baseline)

- Average value of absorbance hover around the range of 80-85, but it shows a distinct pattern how pure water react to different colors of LED.
- Based upon water's nature of its molecular vibrations and electronic transitions, the absorbance suppose to go from low-high (blue→green→yellow→orange→red).

Test Condition (Ta=25°C, IF=20mA)				
	Color	Wavelength (nm)	Light Intensity (med)	Angel
	Red	620-630nm	2000-3000	120°
	Blue	460-465nm	4000-5000	120°
	Green	520-525nm	14000-16000	120°
	Yellow	588-595nm	2000-3000	120°
	Orange	600-605nm	2000-3000	120°

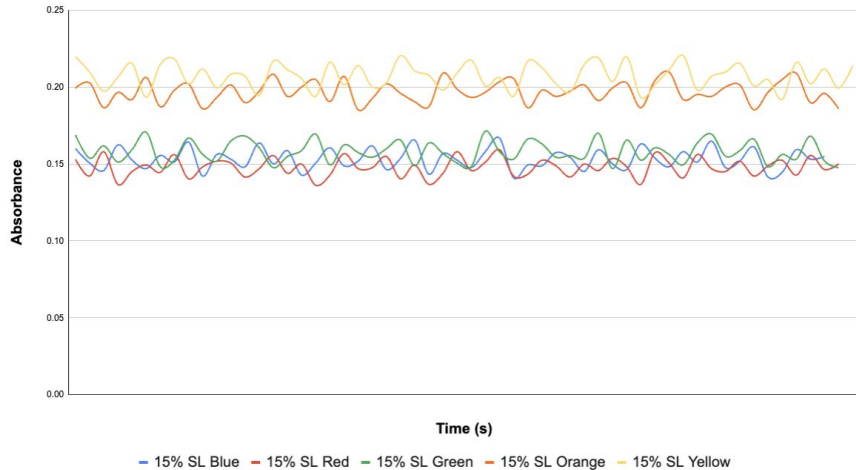
## Pure Water Comparison



- Average value from 4 trials

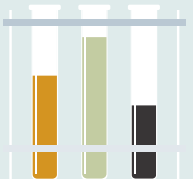
# Data Analysis (15% Saline Water)

## 15% Saline Water



- The absorbance values for 15% saline water are generally lower across all wavelengths compared to pure water.
- **Effect of Ions in Saline Solution ( $\text{Na}^+$ ,  $\text{Cl}^-$ )**
- The presence of dissolved ions increases the overall refractive index of the solution.
- This reduces water's ability to absorb certain wavelengths, leading to slightly higher transmission (lower absorbance).

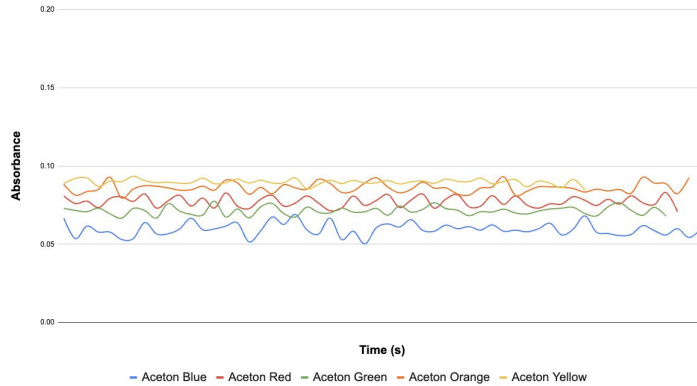
- Average value from 4 trials



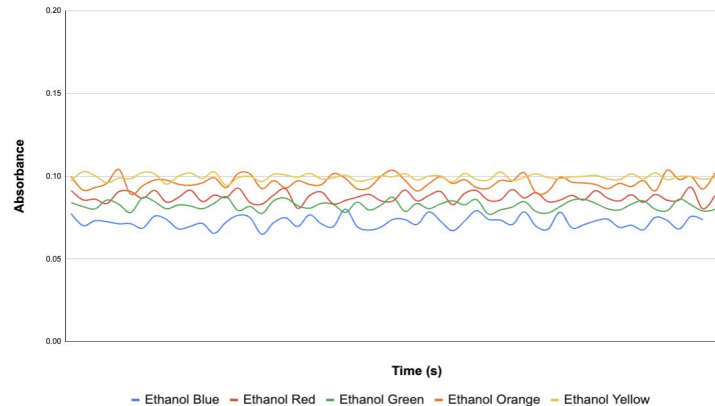
# Data Analysis (Acetone/Ethanol)

- Ethanol and Acetone exhibit higher average absorbance values compared to pure water.
- They both have about the same average absorbance value, which is reasonable for their similar refractive index.
- The molecular and electronic structure affects light transmission by creating different polarity and

Aceton Comparison

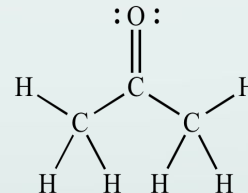


Ethanol Solution

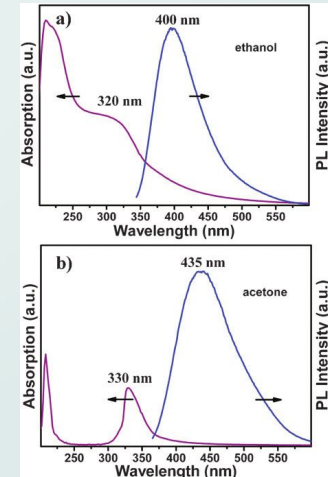
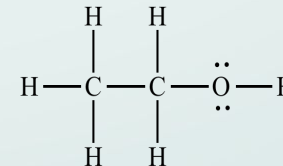


- Average value from 4 trials

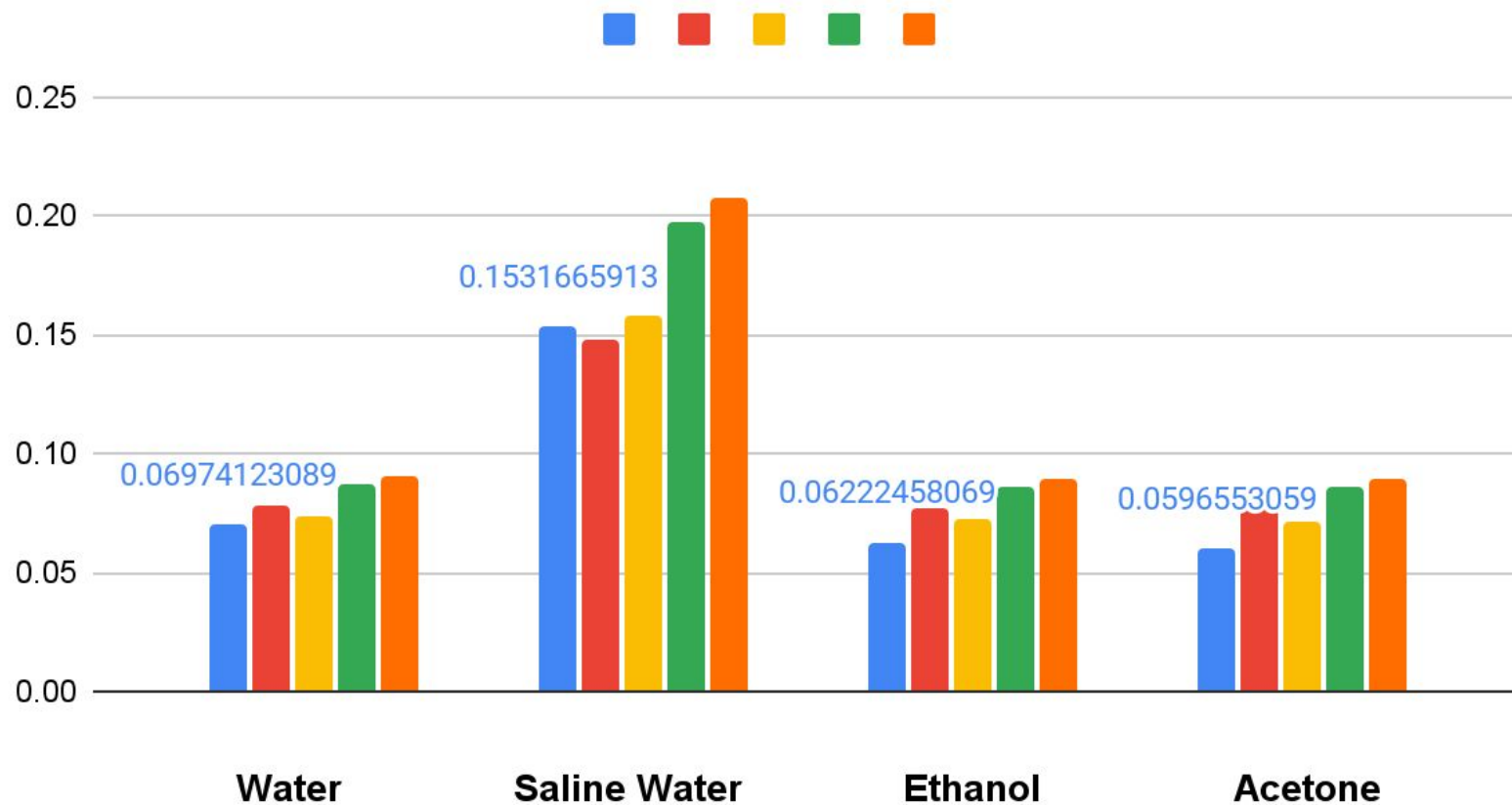
Aceton:  
1.358



Ethanol:  
1.361



## All Solutions' AU Comparison



# Data Comparison

	Pure Water	Saline Water	Ethanol	Acetone
Average AU	0.08	0.17	0.078	0.075
Standard Error	0.03%	0.1%	0.04%	0.04%

# Conclusion

My device did achieve my hypothesis and shows the difference optical density among solutions. However, there are some minor flaws that affect the accuracy. Overall, it's a mimicable and interesting device that can be made easily and understand the concept behind making it.



## References

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